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NUCLEAR CHEMISTRY 701 SECTION 22-1 OBJECTIVES Explain what a nuclide is, and describe the different ways nuclides can be represented Define and relate the terms mass defect and nuclear binding energy Explain the relationship between nucleon number and stability of nuclei Explain why nuclear reactions occur, and know

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Nuclei with 2, 8, 20, 28, 50, or 82 protons or 2, 8, 20, 28, 50, 82, or 126 neutrons tend to be more stable than nuclides with a different number of nucleons. These numbers are called magic numbers . Paired neutrons and protons have a special stability something like the paired electrons have a special stability.

Chapter 21 Nuclear Chemistry

CHAPTER 6 REVIEW Chemical Bonding MODERN CHEMISTRY 10 If concept review answers chap 13, holt modern chemistry answer key ch 17, holt Section 4 3 l a 5 c Chapter 4 Small-Scale Lab Youre The Chemist 1 Chapter 21 profpaz.com ANSWER KEY 1

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REVIEW Use only a periodic table to answer.

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Nuclear Chemistry - Practice ... 718 Chapter 18 Nuclear
Chemistry + + + + Energy $p + p + n + n$ 24 He^{2+} For many of
the lighter elements, the possession of an equal number of
protons and neutrons leads to stable atoms. For example,
carbon-12 atoms, Page 9/28

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Chapter 18 Nuclear Chemistry Answer Key

These lecture presentations were designed for my high school Chemistry I Honors class. Students of high school and college general chemistry may find them useful as a supplement to their own class notes or as a review. Teachers, please feel free to use and modify them for your own classes.

Mrs. J's Chemistry Page - Lecture Notes

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