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problem. But, gone you can support others to begin reading, it will be better.

### **Principles Of Biochemistry Problems And Solutions**

Biochemistry 3300  
Problems (and  
Solutions) Metabolism I  
(11) Place an 'x' on  
each of the high  
energy bonds shown in  
the structures below.

(12) Consider the



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oxidation-reduction reaction between cytochrome b and ubiquinol.

## **Biochemistry 3300**

### **Problems (and Solutions)**

#### **Metabolism I**

1.) In lab, you're having fun with buffers based on a 0.1 M solution of sodium phosphate at pH 7.0. In this system, there are three pKAs: 12.7, 7.21, and 2.12. A. (10 points)

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### Problems And Solutions

You may recall that in the pre-lab lecture on the first day, you were shown a chart indicating the relative amounts of different forms of phosphate ( $\text{H}_3\text{PO}_4$ ,  $\text{H}_2\text{PO}_4^-$ ,  $\text{HPO}_4^{2-}$ ,

## **Chemistry/Biology**

### **302 - Biochemistry:**

#### **Exam 1 Practice**

#### **Problems**

Solutions for Practice Problems for Biochemistry, 6:

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### Problems And Solutions

Glycolysis, Respiration  
and Fermentation

Question 1 ADP ATP  
NAD + NADH

Glycolysis can be summarized as:  
glucose pyruvate a) In a yeast cell, what is the fate of the carbon in pyruvate under anaerobic conditions?  
ethanol

**Solutions for  
Practice Problems  
for Biochemistry, 6**

Solutions to Practice  
*Page 11/24*

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## Biochemistry Problems And Solutions

Problems for  
Biochemistry, Session  
3: Macromolecules:  
Lipids, Carbohydrates,  
Nucleic Acid Question 1  
The following structure  
is found in many  
biological systems.

### **Solutions to Practice Problems for Biochemistry, Session 3**

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In this problem, a  $10^{-10}$  M solution of HCl contributes  $10^{-10}$  M  $[H^+]$ . The ionization of water contributes  $10^{-7}$  M  $[H^+]$ . The effective  $[H^+]$  of this solution is  $10^{-7}$  M and the  $pH=7$ . What is the  $pH$  of a  $10^{-10}$  M ... BioChemistry Problems. let's make good business with this blog. How The Cell Maintain the  $pH$  of Solution in The ...

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Proline does not fit

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well into the structure of the  $\alpha$  helix, because of both its geometry (see Problem 29) and the absence of a peptide  $\rightarrow$ NH to contribute to hydrogen bonding (see Problem 37). This amino acid substitution would produce a protein with decreased stability, which would affect the ability of red blood cells to deform their shape in order to squeeze through

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