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www.collegeboard.org. Question 3 (9
points) $\text{MgO(s)} + 2 \text{H}^+(\text{aq}) \rightarrow \text{Mg}^{2+}(\text{aq}) + \text{H}_2\text{O(l)}$ A student was assigned the task
of determining the enthalpy change for
the reaction between solid MgO and

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GUIDELINES Question 1 (10 points)
Answer the following questions about
the solubility of some fluoride salts of
alkaline earth metals. (a) A student
prepares 100. mL of a saturated solution

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of MgF_2 by adding 0.50 g of solid MgF_2 to 100. mL of distilled water at 25°C and stirring until no more solid dissolves.

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SCORING GUIDELINES Question 1 (10
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Response must clarify each of the
following: o degrees of freedom (df) = 2
and $p = 0.05$ (critical value = 5.99)

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AP[®] Chemistry scores fluctuate slightly, changing by only a few percentage points in recent years. The major exception is between the 2013 and 2014 AP[®] Chemistry Exam. AP[®] Chemistry was redesigned for the 2013-2014 school year, and the exam had significantly fewer 1's, 4's, and 5's and more 2's and 3's.

AP[®] Chemistry Score Calculator for 2020-2021 | Albert.io

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Question 3 (9 points) $\text{MgO(s)} + 2 \text{H}^+(\text{aq}) \rightarrow \text{Mg}^{2+}(\text{aq}) + \text{H}_2\text{O(l)}$ A student was assigned the task of determining the enthalpy change for the reaction between solid MgO and aqueous HCl represented by the net-ionic equation above.

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Chief Reader Report on Student Responses - AP Central

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